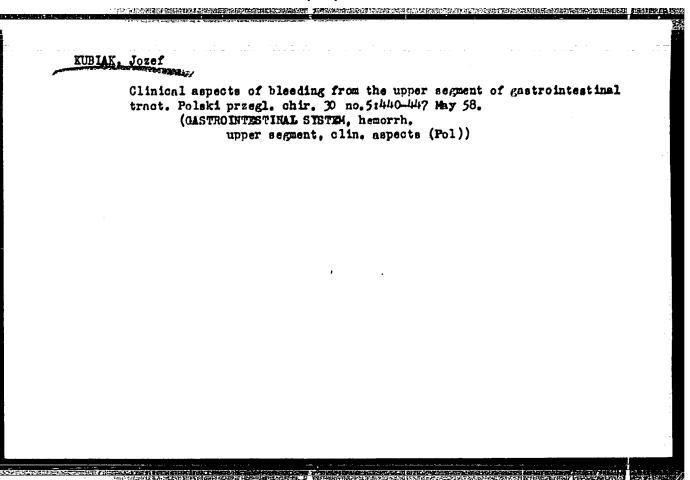
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KUBIAK, Josef; ZAMBCKI, Stanislaw; LISZKA, Wladyslaw

Late complications following surgery of the biliary tract. Polski przegl. chir. 29 no.5:439-446 May 57.

1. Z Oddzialu Chirurgicznego Szpitala na 8 w Warszawie Kierownik: doc. dr J. Kubiak Prace Wplynela dnia 27. VI. 1956 r. Warszawa, ul. Solec 93.

(BILIARY TRACT, surgery, postop. compl. (Pol))



KUBIAK, Jozef; BADMAJEW, Piotr

Pyloric hypertrophy in adults. Pol. przegl. chir. 34 no.12:1265-1270 162.

1. Z I Kliniki Chirurgicznej SDL w Warszawie Kierownik: doc. dr J. Kubiak.

(PYLORIC STENOSIS)

DESIGNATION OF THE PROPERTY OF

### KUBIAK, M.

The solar magnetic field and the 22-year cycle. Postepy astronom 10 no.3:287-292 '62.

公。2006年的经验结婚在西面中的经验的特别被保持的特别的特别的经验的企业,在全国的特别的企业的企业,从2007年,1951年的经济的企业的企业的发展的专用的基础的**是,但是国际政策等的** 

# New observations of the red-shift of solar lines. Postepy astronom 11 no.4:289-293 '63.

### KUBIAK, Marian

Preliminary results of research on impregnation treatments of standing trees. Sylvan 56 no.1;73-77 Ja-F 162.

KUBIAK, Marian; ROGALINSKI, Kamil; MICHALAK, Jermy

Changes in the physical and mechanical properties of poplar wood (Populus marilandica Bosc.) under the influence of Poria vaporaria (Pers/Fr.) fungus. Sylvan 106 no.3:25-39

'62.

KUBIAK, Marian; KERNER, Gunter

Changes of some physical properties, compression strength and chemical composition of beechwood in the initial stage of decomposition by fungi Coniophora cerebella Pers. and Stereum hirsutum Willd. Drevarsky vyskum no.4:181-193 163.

1. Institut fur Forstnutzung, Landwirtschaftliche Hochschule, Poznan (for Kubiak). 2. Institut fur physikalische Holztechnologie und Chemisches Institut fur Forstvirtschaft, Sberewalde (for Kerner).

。 第111章 "我们的一个人,我们就是我们的一个人,我们们的一个人,我们们就是我们的人,我们们就是我们的人,我们就是这些人,我们就是一个人,我们就是一个人,我们就

### KUBIAK, Marian

Studies on the possibilities of demarcating the types of wood rot on the basis of the physical, mechanical, and chemical properties caused by fungi. Roczniki wyz szkola rol Poznan 18 57-90 163.

 Department of Forest Utilization, College of Agriculture, Poznan.

THE STANDARD STANDARD STANDARD STANDARD STANDARD STANDARD STANDARD STANDARD AND STANDARD STAN

KUBIAK, Marian, dr 🚬

Influence of fungi on the changes in the chemical composition of beechwood (Fagus silvetica). Przegl papier 20 no. 1: 7-11 Ja 164.

1. Wyzsza Szkola Relnicza, Poznan, Katedra Uzytkowania Lasu.

TO THE TRANSPORT THE PROPERTY OF THE PROPERTY

KUBIAK, Marian

Research on the effect of various fungi on the change of technical properties in the wood of Populus marilandica. Bosc. Drevarsky vyskum no.42171-185 164.

1. Chair of Forestry, Higher Fonool of Agriculture, Poznan, Poland.

KUBIAK, Marian, dr

Influence of fungi on the changes in the chemical composition of poplar wood. Przegl papier 20 no.7:209-213 Jl '64

1. Department of Forest Utilization, School of Agriculture, Poznan.

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AP6033312

SOURCE CODE: PO/0009/66/016/002/0151/0155

PRESENTATION DE PARTICION DE LA COMPANSION DEL COMPANSION DE LA COMPANSION DE LA COMPANSION DE LA COMPANSION

AUTHOR: Kubiak, M.; Stepien, K.

ORG: Warsaw Astronomical Observatory

TITLE: Some remarks on ionization fronts in interstellar matter containing

molecular hydrogen

SOURCE: Acta astronomica, v. 16, no. 2, 1966, 151-155

TOPIC TAGS: interstellar matter, hot star, molecular hydrogen, ionization,

ionization front

ABSTRACT: A short discussion is presented of atomic processes which might be important in an ionization front passing through an interstellar cloud containing molecular hydrogen. All these processes are shown to occur simultaneously with no separation between ionization and dissociation fronts. It is shown that the addition of molecular hydrogen does not modify substantially the dynamic effects of the interaction of interstellar gas and the radiation field of hot stars. One front can exist in interstellar gas and its structure is determined mainly by processes two and five described in the original article. Orig. art. has: 2 figures, 5 formulas and table card SUB CODE: 03,20/SUBM DATE: 00Nov65/OTH REF: 005/

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EUTAK, M.

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Periodicals: 0920 LAD MEC GIOZNY Vol. 79, No. 19, October 1958

FU TAK, 1. Some problems of the use and saving of noncerrous metals in the years 1959-1965. P. 8:9.

Monthly List of East European Accessions (REAI) LC, Vol. 8, No. 2, February 1959, Unclass.

KUBIAK, M.; ZIENTARA, J.

Proper use of nonferrous metal scrap in industrial enterprises. p. 9.

PRZEGIAD TECHNICZNY. (Naczelna Organizacja Techniczna) Warszawa, Poland Vol. 80, no. 17, April, 1959.

Monthly List of East European Acessions Index, (EEAI), IC, Vol. 8, no. 64. June 1959 Uncl.

KUBIAK, M.; ZIENTARA, J.

Plastics as substitute materials for nonferrous metals. p. 1.

FRZEMLAD TECHTICZNY. (Naczelna Organizacja Techniczna) Warszawa, Foland. Vol. 80, no. 23, June 1959.

Monthly List of East European Accessions (FEAI), LC, Vol. 8, no. 8, Aug. 1959. Uncl.

### KUBIAK, M.

Technical Progress and increased economy will bring about a saving of copper. p. j. PRZEGLAD TECHNICZNY. Naczelna Organizacja Techniczna. Warszawa, Poland, Vol. 80, ro. 28, July 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 9, September, 1959. Uncl.

KUBIAK, M.: ZIENTARA, J.

Plastics instead of lead. p.7

FRZEOLAD TECHNICZNY. (Nacselna Organizacja Techniczna) Warszawa, Poland Vol.80, no.41, Oct. 1959

Monthly list of East European Accessions (EEAI) LC, Vol.9, no.1, Jan. 1960 Unel.

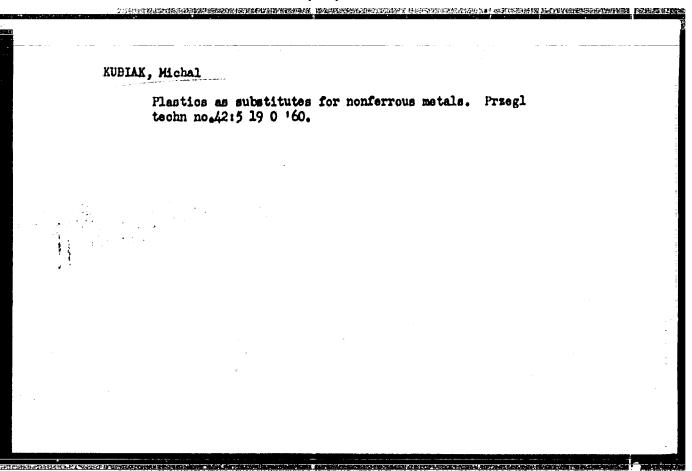
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### KUBIAK, M.

Saving nonferrous metals. p.3

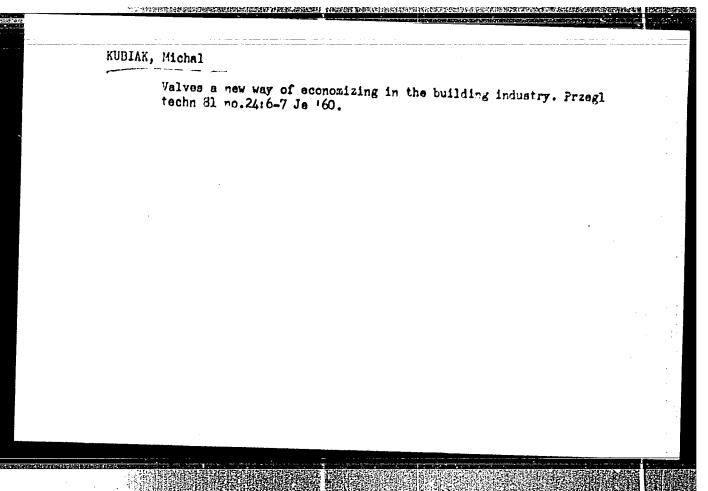
PRZECLAD TECHNICZNY. (Naczelan Organizacja Techniczna) Warszawa, Poland Vol.80, no.47, Nov. 1959

Monthly list of East European Accessions (EFAI) LC, Vol.9, no.1, Jan.1960 Uncl.



KUBIAK, Michal

The electric engineering industry economises on nonferrous metals. Preegl techn no.43:5 26 0 160.



KUBIAK, Michal

Technical progress in the application of aluminum during the years 1961 - 1965. Przegl techn 81 no.25: 10-12 Je '60.

GRYCUK, T., KUBIAK, M., PROCHOROW, J.

A PARTICULAR PROPERTY OF THE P

Temperature effect on pressure broadening of the mercury resonance line. Bul Ac Pol math 12 no.8:517-522 664.

1. Institute of Experimental Physics of the University, Warsaw. Presented by A. Jablonski.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827010020-1"

S/081/62/000/001/024/067 B151/B101

AUTHORS:

Kreshkov, A. P., M., Drozdov, V. A., Vlasova, Ye. G., Kubiak, S.

TITLE:

Determination of organosilicon compounds by titration in a non-aqueous medium

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 1, 1962, 158, abstract 1D158 (Vestn. tekhn. i ekon. inform. N.-i. in-t tekhn.-ekon. issled. Gos. kom-ta Sov. Min. SSSR po khimii, no. 10, 1960, 29-32)

TEXT: Methods are described for the analysis of alkyl(aryl) chlorosilanes (ACS), alkyl(aryl) (alkoxy)sminosilanes (AAS) and silamines (SA), based on their titration in non-aqueous media. It is shown that titration of non-aqueous solutions of these compounds can be carried out using indicators, potentiometry, conductivity measurement and high frequency methods. ACS are titrated in a medium consisting of mixed solvents;  $CH_2CN - C_6H_6$  (1 : 1) with 0.1 - 0.05 % acetonitrilic solutions of nitron (diphenylendanilohydrotriamals)

Card 1/2

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Determination of organosilicon ...

(I) and pyridine (II) or 0.1 - 0.05 M benzene solution of dimethylaminoantipyrine (III) in the presence of the usual indicators (crystal violet, dimethyl amino azo benzene, bromocresol purple etc.). The titer of solution I is determined using an accurately weighed sample, while that of solution II is determined using HClO4. The best results are obtained by titrating with solution III. With potentiometric determination the ACS is titrated with solution III using glass and calomel electrodes. The error of the method is ± 0.5%. Conductometric determination gives the best results by titrating the ACS with 0.1: M benzene solution of III; error of the method & 0.5%. The differential conductometric titration of a mixture of methylchlorosilanes (MCS) is based on a preliminary quantitative conversion of the MCS by the action of NHASCN into methylthiocyanate substituted products (MTS) and subsequent conductometric titration of the MTS with solution III in a medium consisting of acetonitrile and diethyl ether. [Abstracter's note: Complete translation.

Card 2/2

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158312

Kreshkov, A.P., Drozdov, V.A., and Kubiak, S.

TITLE:

AU THORS:

Properties of alkylchlorosilanes in acetonitrile medium

PERIODICAL: Zhurnal obshchey khimii, v. 31, no. 9, 1961, 3099 - 3103

TEXT: The present work deals with the conductometric examinations of methylchlorosilanes. The conductometric titration was carried out in a series of organic solvents and the measurements were taken of specific and equivalent conductivities of methylchlorosilanes in acetonitrile medium. The choice of acetonitrile was made due to its excellent properties as a solvent for inorganic, organic and organosilicon compounds, ability to increase the dissociation of salts, acids and bases and also its low association in liquid state as compared with that of methanol or ethanol. According to R.S. Mulliken (Ref. 4: J. Phys. Chem. 56, 814, 1952) donoracceptor classification acetonitrile may function in two ways, as base conclassification acetonitrile may function in two ways, as base con-

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taining easily ionizable pair of electrons and as  $\pi$  -ketoidal acceptor. The compounds used in the investigations were pure trimethylchlorosilane, dimethyldichlorosilane and methyltrichlorosilane, titrants were organic bases containing tertiary nitrogen atoms (pyridine, quinoline, 8-hydroxyquinoline, dimethylaminoantipyrine etc.). Conductometric titration was conducted with the aid of kohlrausch bridge and a closed-type cell with sealed platinum plated platinum electrode. Measurements of electroconductivity of methylchlorosilanes in acetonitrite medium was carried out in thermostatically controlled 50 cc spherical vessel. The coice of medium for conducting the measurements was carried out experimentally using methylchlorosilane in various organic solvents such as acetonitrile, benzonitrile, nitromethane, nitrobenzene, cyclohexanone, methyl ethyl ketone and methyl butyl ketone, and an organic base as a titrant. Typical conductometric titration curves are given in Fig. 1, and it is shown that the best medium is provided by acetonitrile. The titration curves for various methylchlorosilanes are given in Fig. 2. They are shown to exhibit two characteri-

Card 2/7

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Properties of ...

stic points: one (minimum) which corresponds to about half-way titration, and the other (beginning of horizontal portion) exact end point also given by bromocresol purple indicator (colorless to bright yellow). On reacting with organic bases, methylchlorosilanes behave as acids and the chemical processes may be represented as follows:  $2R_nSiCl_{4-n}$  + amine  $\rightarrow$   $(2R_nSiCl_{4-n})$  amine for the half -way point and for the end point:  $(2R_n \operatorname{SiCl}_{4-n})$  amine + amine  $\implies$  2 [(RnSiCl4-n) amine]. The formation of this complex is accompanied by the increased conductivity of the solution due probably to its dissociation. On the other hand the half-way titration complex is more stable and causes the increase in the resistance of the solution accompanied by the reduction in conductivity. To establish the possible electrolytic dissociation of pure methylchlorosilanes the authors have measured conductivity of the latter in acetonitrile medium at constant temperature (26°C). The variations of specific and equivalent conductivities of methylchlorosilanes on dilution with acetonitrile solutions is represented in Fig. 4. As Card 3/7

Properties of ...

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the dilution increases  $\lambda$  increases as a result of higher degree of dissociation (a). The relationship of  $\lambda$  to dilution shows that methylchlorosilanes behave as weak electrolytes in non aqueous media and conform to Ostwald's law. The process of formation of currentconducting acetonitrile solutions of alkylchlorosilanes may be illustrated on the example of trimethylchlorosilane. Knowing the ionic mobility of the dissociation products it is possible to calculate dissociation constants for every methylchlorosilane. The degree of dissociation is highest for trimethylchlorosilane and lowest for methyltrichlorosilane. There are 4 figures and 6 references: 2 Soviet-bloc and 4 non-Soviet-bloc. The 3 references to the English-language publications read as follows: E.A. Abrahamson, C.A. Reynolds, Anal. Ch., 24, 1827, 1952; R.S. Mulliken, J. Phys. Chem., 56, 814, 1952; E.A. Jerger, G.M. Barrow, J. Am. Chem. Soc., 77, 4474, 6-06, 1955. ASSOCIATION: Moskovskiy khimiko-tekhnologicheskiy institut im. D.I.

Mendeleyeva (Moscow Institute of Chemistry and Tech-

nology im. D.I. Mendeleyev) SUBMITTED: September 30, 1960

Card 4/7

THE PROPERTY OF THE PROPERTY O

KOZLOWSKI, L.; KUBIAK, S.

Effect of cathodic hydrogen on the magnetic properties of thin electrolytic nickel films. Bul Ac Pol mat 9 no.5:409-417 '61.

1. Department of Physics, Normal School, Katowice. Presented by M. Miesowicz.

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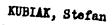
KOZLOWSKI, L.; KUBIAK, S.

Changes in the magnetic moment of thin electrolytic nickel films cathodically polarized. Bul Ac Pol mat 11 no.4:235-

1. Department of Physics I, School of Mining and Metallurgy, Krakow, and Department of Physics, Normal School, Katowice. Presented by M. Miesowicz.

KUPRYK, Loch, mgr inz.; KUBIAK, Stefan, inz.

Survey of production of the Electric Lamp Assembly Equipment Works. Wiad elektrotechn 30 no.5:170-172 My '62.



The secret of success in racing on Polish cinders; driving tactics on cinder tracks. Motor 11 no.28:13 15 Jl 162.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827010020-1"

KUBIAK, Stefan

Great Poland is on the route of international touring; interview with Jofez Niewiedzial, secretary of the Polish Motorists Union. Motor 12 no.3:13 20 Ja '63.

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827010020-1"

POLAND / Cultivated Plants. Medicinal Plants. M-9
Essential Oil Plants. Poisonous Plants.

Aus Jour: Ref Zhur-Biol., 1958, No 16, 73224.

Author : Gatty-Kostyla, Marek; Kublak, Z.; Kostolowska, M.

Inst : Not glven.

Title : Medicinal Use of Shoots of Belladonna Instead of

the Leaves.

Orig Pub: Acta polon, pharmac., 1956, 13, No 2, 31-88.

Abstract: It is recommended to gather the tops of the belladonna shoots above the first fork, which contain a great quantity of active compounds, before flowering and during setting of the fruit. The alkaloids content in the shoots comprises 0.4-0.5% which exceeds the normal of the Polish pharmacopoeia 1.5-fold. The greatest content of alkaloids is in the raw material of the spring harvest. Comparison of

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APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827010020-1
POLAND / Cultivated Plants. Redicinal Plants.

Essential Oil Plants. Poisonous Plants.

Abs Jour: Ref Zhur-Biol., 1958, No 16, 73224.

Abstract: productivity of shoots and leaves of belladonna, conducted on an experimental plot, showed that harvest of shoots give 3 times more alkaloids than harvests of the leaves. Harvest of shoots is 3-4 times faster than harvest of leaves. It is proposed to include the shoots of Belladonna in the Polish pharmacopoeia instead of the leaves. -- Yu. I. Rafes.

COUNTRY : Poland H-17 CATEGORY :

ABS. JOUR. : AZKhim., No. 16 1959, No. 58075

AUTHOR : Eublak, 7.

TROY. : Polish Academy of Sciences

TITLE : Improvements in the Method for Preparing Bella-

donna Extracts. Part II. Liquid Extracts.

ORIG. PUB. : Dogissert Pharmac PAN, 10, No 1, 29-38 (1958)

ABSTRACT : The investigation of methods for the preparation

of liquid extracts has shown that the liquid extract obtained by the method described in volume III of the Polish Pharmacopeia represents a stable and therapeutically valuble preparation. A simplified method for the preparation of liquid extracts has also been developed, and the practical value of their application has been demonstrated. For Part I are Polyholas 1058. To 16

strated. For Part I see RZhKhim, 1958, No 16,

54925.

From author's summary

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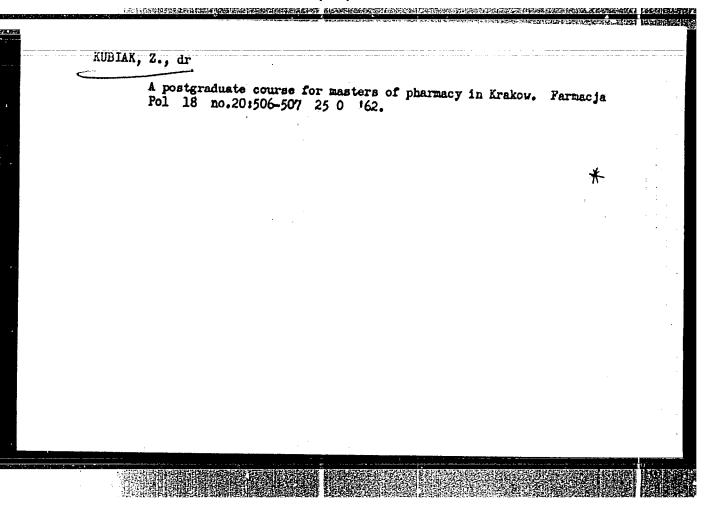
### KUBIAK, Z.

Texic compounds of tebacco smeke. P 105

WSZECHSWIAT. (Polskie Towarzystwe Przyrodnikow im. Kepernika) Warszawa. R. A. No. 4, Apr. 1959

Monthly List of East European Accessions (EEAI) LC. Vol. 9, no. 7, July 1959

Uncl.



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DOMANSKA, H., mgr; KUBIAK, Z., dr, adiunkt

Hospital pharmacies in Krakow. Farmacja Pol 19 no.3:53-55 10 F '63.

1. Kierownik Oddzialu Farmacji Wydzialu Zdrowia i Opieki Spolecznej, Powiatowa Rada Narodowa, Krakow (for Domanska). 2. Katedra Farmacji Stosowanej, Akademia Medyczna, Krakow (for Kubiak).

X

KUBIAK, Z.

From the activities of the Hospital Pharmacy Section of the Krakow Branch of the Polish Pharmaceutical Society. Farmacja Pol 19 no. 15/16:350 25 Ag '63.

KUBIAK, Z., dr.

From the Section of Hospital Pharmacies of the Krakow Branch

of the Polish Pharmaceutical Society. Farmacja Pol 20 no.1/2: 61 25 Ja 64.

KUBIAK, Z., dr.

Supplementary course for Masters of Pharmacy in Krakow. Far-macja Pol. 19 no.17/18:377-378 25 S'63

From the Krakow Branch of the Polish Pharmacoutical Society. Ibid.:384.

From the Analytical Section of the Krakov Branch of the Polish Pharmaceutical Society. Tbid.:385

TO THE PARTY OF TH

RUBLES, Shigniew

Concentration of grain alcohol and the composition of active substances in liquid extracts from the bark of the common buckthorn. Farmacja Pol. 20 no. 5/6:176-181. 25 Mm 364.

From the Cention of Mospital Thornworks of the Frakow branch of the Polish Pharmaceutical Society, Thin.:213

1. Department of Applied Pharmacy School of Nericine, Krakowe Head: Frof. dr M.Gatty-Kostyal.

Kutens JAREMIZ

Czechoslovakia Chemical Technology. Chemical Products 1-27 and Their Application

Wood chemistry products. Cellulose and its manufacture. Paper.

Abs Jour: Referat Zhur - Khimiya, No 9, 1957, 32721

Author : Kubias Jaromir

Title : Determination of the Density of Paper Pulp with

a Pencil Densimeter

Orig Pub: Papir a celulosa, 1956, 11, No 7, 163-164

Abstract: The pencil densimeter consists of an aluminum

rod of precise dimensions, pointed at one end and previously calibrated. Paper pulp at 20-25° is placed in a measuring cylinder. The densimeter drops into the pulp due to its own weight from a definite height (2.5 cm). From

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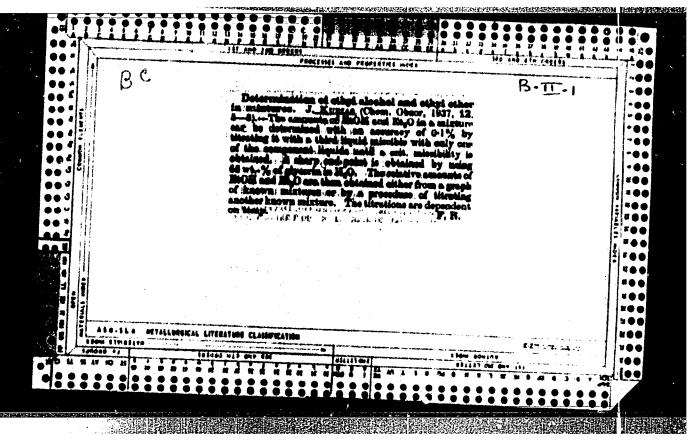
Czechoslovakia Chemical Technology. Chemical Products I-27 and Their Application

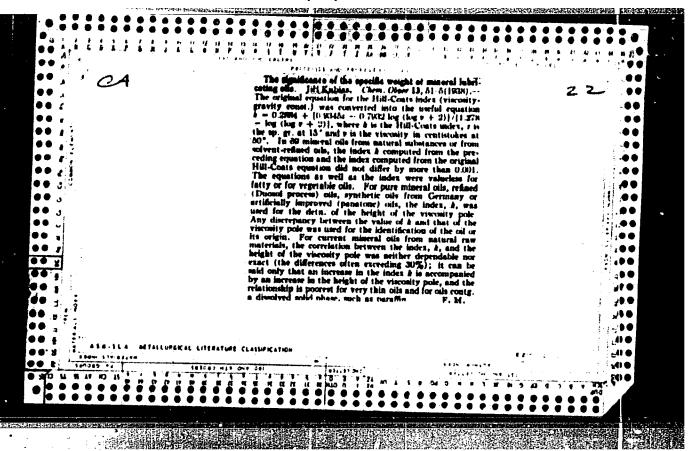
Wood chemistry products. Cellulose and its manufacture. Paper.

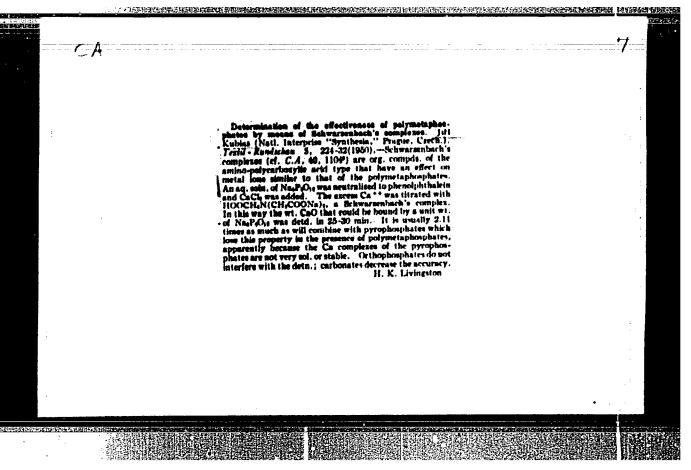
Abs Jour: Referat Zhur - Khimiya, No 9, 1957, 32721

the depth of its penetration into the pulp the density of the pulp is ascertained. Determination error is 0.5%. The advantage of the measurements.

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KUBIAS, J.

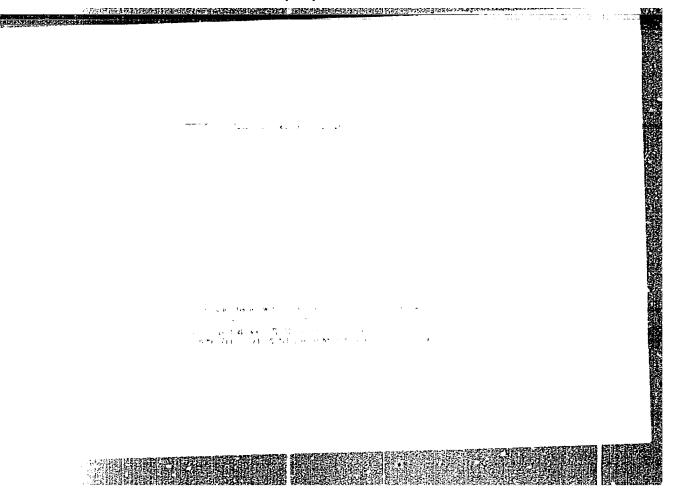
Kubias, J.; Fleissig, O. "A Quick and Precise Method of Determining the Hardness of Water." p. 91 (Uhli A Rudy, Vol. 1, no. 4, Apr. 1951, Praha)

So: Monthly List of Russian Accessions,/Library of Congress, March 1954

KUBIAS, J., PILMY, S.

"Determination of Grotonaldehyde in Butyraldehyde by Direct Potenticmetric Titration with Bromine" p. 672., (CHEHICKE LISTY, Vol. 47, no. 5, May 1953, Praha, Czechoslovakia).

SO: Monthly List of East European Accessions, LC, Vol. 2, No. 11, Nov. 1953, Uncl.



APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000827010020-1"

CZECHOSLOVAKIA / Analytical Chemistry. Organic Analysis. = KUBIAS. : Ref Zhur - Khimiya, No 23, 1959, No. 82041 Abs Jour : Kubies, J. : Not given : The Determination of Ethanolamines in Their Author Inst Title Mixture : Collect. Czechosl. Chem. Communs., 1958, 23, Orig Pub No 11, 2012-2017 : See RZ Khim, No 3, 1959, No. 8038 Abstract card 1/1 37 

### KUBIAS, M.

Simple colloidal cerebrospinal reaction. Cas. lek. cesk. 90 no.31:936-940 3 Aug 1951. (CIML 21:1)

1. Of the Cerebrospinal fluid Laboratory of the Psychiatric Clinic in Prague (Head -- J. Prokop, M.D.) and of the Neurological Clinic of the Medical Faculty in Pilsen (Head -- Prof. J. Hrbek, M.D.).

PIRK, P.; BELAN, A.; TRAVNICEK, R.; BUDINOVA-SMELA, J.; FRYNTOVA, A.: technicke spoluprace BUPKA, L.; KRIZOVE, M.; KUBIASOVE, E.; KUTILA, L.

Our experiences with roentgen cinema tography in cerebral angiography. Preliminary report, Cesk. neur. 24 no.1:51-53 Ja 161.

1. Ustav pro vyskum vysivy lidu, Praha, reditel doc. MUDr. J. Masek - Ustav pro klinickou a experimentalni chirurgii, Praha, reditel profesor MUDr. B. Spacek - Oddeleni pro cevni onemocneni mosku, predn. doc. MUDr. J. Budinova-Smela, Laboratore statniho filmu, Barrandov.

(CEREBRAL ANGIOGRAPHY)

SCHINDLERY, C.; Nadvornik, P.; Nemecek, S.; Kusikovi, E.

Stereotaxic model of cerebellar nuclei. Cosk. neurol. 27
no.6:370-371 N \*64.

1. Neuroccirurgicka klinika lekarske fakulty Karlovy University v Hradel Kralove, (prednosta prof. dr. E. Petr.).

### EUBIATCH HI, J.

"Vortilation of Trunoformers! Barke", P. 250, (MIANCMOSCI TELCIRCH CISH ZME, Vol. 14, No. 11, November 1954, Warsaw, Poland)

SC: Venthly List of Fast European Accessions (MAL), 10, Vel. 4, No. 3, March 1955, Uncl.

The Commission for Utilization of Wind Power of the Folish Slectrical Engineers Association. p. 94, Vol. 15, no. 4, Apr 1955, WIADOMOSCI ELECTROTECHNICZNE

JOHN NTHLY LIST OF EAST EUROPEAN ACCESSIONS, (EEAL), LC, Vol. 4, No. 9, Sept. 1955, Uncl.

KUBIATOW WI, J.

Studies on the standardization of electric lighting in the wool-using industries. p.278 THREALIMOJA (Polski Komitet Mervalimosjay) Marasewa Vol. 23, no. 5, May 1955

So. Mast European Accessions List Vol. 5, Mo. 1

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CIA-RDP86-00513R000827010020-1" **APPROVED FOR RELEASE: 03/13/2001** 

Kubiatowski, J. Possibilities of applying air turbines in forestry. p. 34

LOS FOLSKI

Vol. 29, No. 6, June 1956 Warszawa, Poland

SO: Monthly List of East European Accessions, (EEAL), IC, Vol. 5, No. 10 Oct. 56

CIA-RDP86-00513R000827010020-1" **APPROVED FOR RELEASE: 03/13/2001** 

"New regulations on installing low-voltage electric power lines along highways."

p. 236 (Wiadomosci Elektrotechniczne) Vol. 17, no. 9, Sept. 1957 Warsaw, Poland

50: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4, April 1958

"Wind-motor electric plants with reserve diesel engines."

p. 237 (Wiadomosci Elektrotechniczne) Vol. 17, no. 9, Sept. 1957 Warsaw, Poland

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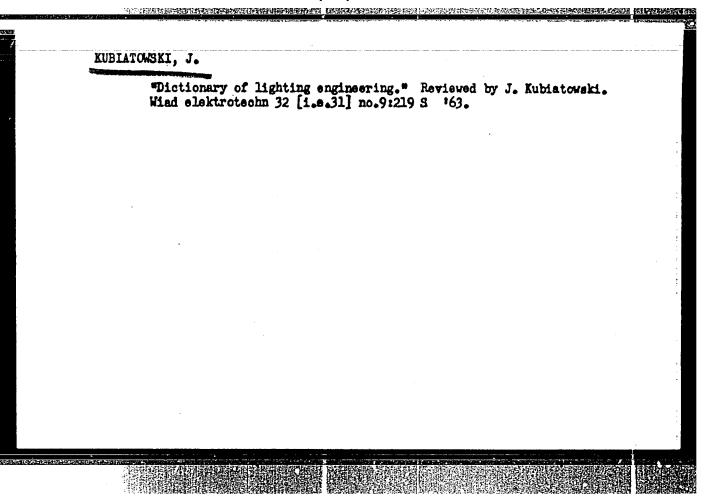
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AUTHORS:

Ziemkiewicz, J., Kubica, J., Maik, J., Mostek, L.

TITLE:

Method of bonding aluminum and its alloys with ferrous alloys

PERIODICAL: Referativnyy zhurnal. Metallurgiya, no. 9, 1961, 20, abstract 9D143P (Fabryka Samochodow Gsobowych w Warszawie, Polish patent no, 43225,

TEXT: To obtain a high grade aluminum coating on cast iron, carbon or alloy steel, and also other Fe compounds after a thorough cleaning, chemical degreasing and anode cleansing, a film of electrolytic Cu with thickness 1-2  $\mu$  is produced in a vat with composition (in grams per liter): NaCN 30, CuCn 60 in the course of 2 - 25 min. Thereupon the part is slowly immersed for 15 min in melted Al having a temperature of 690 - 720°C; this produces a diffusion layer with thickness 15 - 25 pon the surface of the part. The part is taken out of the vat and not later than within 30 sec it is placed in a foundry mold which is fleoded with Al or its alloy. The obtained Al-Fe compound has o'b 7.76 kg/mm and possesses great hardness, durability, thermal conductivity, heat capacity, corrosion

Card 1/2

Method of bonding aluminum ...

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resistance, low specific gravity, and may be used, for example, for the manufacture of cylinders of engines with water or air cooling, or of brake drums.

V. Levinson

[Abstracter's note: Complete translation]

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Card 2/2

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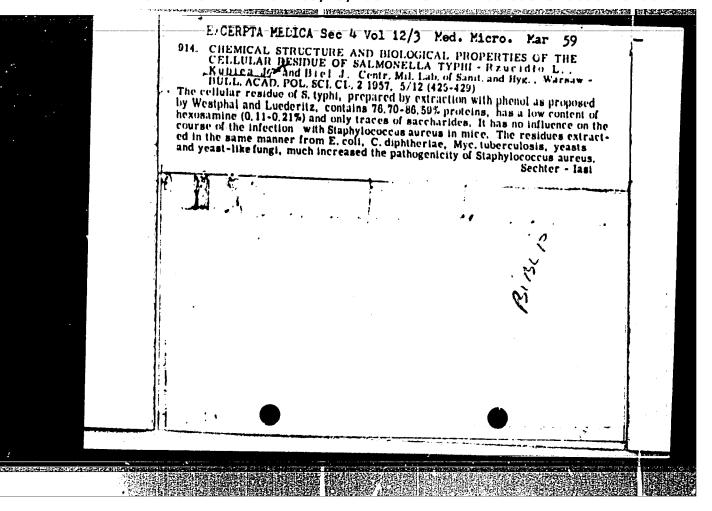
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(PLASTICS) (POLYETHYLENES) (POLYVINYLS) (INFUSIONS, PARENTERAL) (INJECTIONS)

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JIRU, Pavel; KUBICA, Ludvik

Stabilized vanadium oxidation catalysts for sulphuric acid production. Chem prum 12 no.5:232-236. My 162.

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